

Highly Regio- and Stereoselective Synthesis of 2(*E*),4-Alkadienoates via the Pd(0)-Catalyzed Reaction of Aryl Halides with 3,4-Alkadienoates

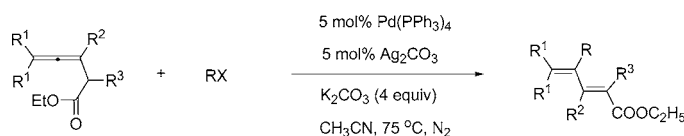
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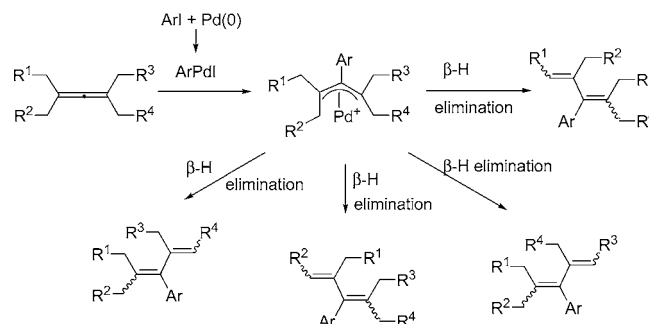
ABSTRACT



2(*E*),4-Alkadienoates were prepared highly stereoselectively via the Pd(0)/Ag₂CO₃-cocatalyzed reaction of 3,4-alkadienoates and aryl halides. The reaction is believed to proceed via the oxidative addition–carbopalladation– β -H elimination process. Compared to the other reported methods for the synthesis of 2,4-alkadienoates, in which usually only disubstituted C=C bonds were formed, the current reaction forms the trisubstituted or even tetrasubstituted C=C bond highly stereoselectively.

Stereoselective synthesis of conjugated 1,3-dienes is of current interest¹ since they are very important intermediates in organic synthesis.² Recently, much attention has been paid to the transition metal-catalyzed chemistry of allenes.^{3–6} After observing high selectivities with allenes bearing an α -functionality,⁴ we proposed that carbopalladation of allenes would provide a very convenient entry to π -allyl palladium species, which upon β -H elimination would lead to a convenient formation of conjugated 1,3-dienes (Scheme 1).⁷ However, the stereoselectivity of β -H elimination is usually disappointing, which makes this approach unsuitable for the highly stereoselective synthesis of 1,3-dienes.⁸ In this paper, we wish to report the first highly regio- and stereoselective Pd(0)-catalyzed reaction of 3,4-alkadienoates with organic halides leading to 2(*E*),4-alkadienoates in high yields. Compared to the other reported protocols,¹ in which usually only disubstituted C=C bonds were formed, the current reaction forms

Scheme 1. Regio- and Stereochemical Issues in the Pd-Catalyzed Synthesis of 1,3-Dienes from Allenes and Organic Halides



the trisubstituted or even tetrasubstituted C=C bond highly stereoselectively.

At the beginning, we tried the reaction of ethyl 3-butyl-3,4-pentadienoate **1a** with phenyl iodide under the conditions for the coupling cyclization of organic halides with 2,3-

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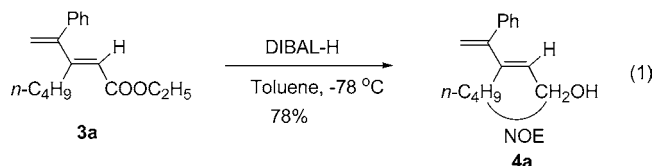
allenoic acids,⁹ which afforded ethyl 4-phenyl-3-butyl-2(*E*),4-pentadienoate **3a** in 80% yield as the only product (entry 5, Table 1). The stereoselectivity is excellent, i.e., only the 2(*E*)-

Table 1. Pd(0)/Ag⁺-Catalyzed Carbopalladation and β -H Elimination Reaction of **1a** with PhI

entry	solvent	temp (°C)	time (h)	PhI (equiv)	yield of 3a (%) ^a
1	toluene	75	13	1.5	10
2	THF	75	14	1.5	45
3	dioxane	75	14	1.5	65
4	DMF	75	14	1.5	76
5	CH ₃ CN	75	13	1.5	80
6	CH ₃ CN	75 ^b	13	1.5	63
7	CH ₃ CN	60	13	1.5	16
8	CH ₃ CN	75	13	1.2	63
9	CH ₃ CN	75 ^c	13	1.5	71

^a Isolated yield. ^b Reaction was conducted in the absence of Ag₂CO₃. ^c Reaction was conducted in the presence of 2 equiv of K₂CO₃.

stereoisomer was formed, which is quite unique as compared to a similar nonstereoselective reaction with normal alkyl-substituted allenes.⁸ The stereochemistry of **3a** was established via the combined NOE study of **3a** and **4a** (eq 1).



After the reaction was run in different solvents, it was observed that MeCN is the best solvent (entry 5, Table 1). Further study indicates the following: (1) The presence of Ag₂CO₃ is necessary since the reaction in the absence of 5 mol % Ag₂CO₃ only afforded **3a** in 63% yield (compare entry 5 with entry 6, Table 1). This may be explained by the interaction between the carbonyl group and Ag⁺ to ensure a fast and regioselective β -H elimination. (2) The reaction temperature is also crucial since the reaction at 60 °C afforded **3a** in only 16% yield (entry 7, Table 1). (3) For a high-yielding reaction, 1.5 equiv of PhI and 4 equiv of K₂CO₃ are required (compare entry 5 with entries 8, 9, Table 1).

With these results in hand, we studied the scope of this transformation; some typical results are summarized in Table 2. From Table 2, it can be concluded that the reaction is general: R² can be H, alkyl, or aryl; both electron-withdrawing and electron-donating aryl halides can be used in this reaction (entries 7–12, Table 2). The reaction with thienyl bromide also proceeded smoothly, affording the

corresponding 2(*E*),4-pentadienoate **3m** in 66% yield (entry 13, Table 2). No β -H elimination toward R¹ or R² was observed. However, under the conditions reported by Cheng et al.,⁸ the reaction afforded a mixture of 1,3-dienes formed from the nonregioselective β -H elimination.

On the basis of the analysis of ¹H NMR spectra of the crude products, the extent of the formation of the 2(*Z*)-isomer

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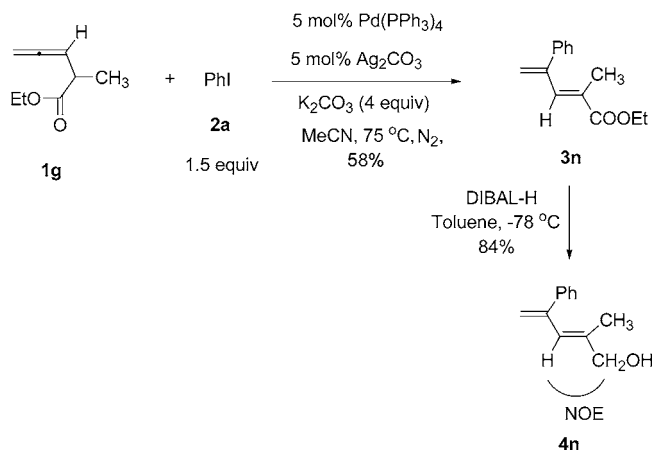
Table 2. Pd(0)/Ag⁺-Catalyzed Reaction of 3,4-Alkadienoates with Organic Halides^a

entry	R ¹	R ²	RX	2/1 (equiv)	time (h)	yield of 3 (%) ^b
1	H	<i>n</i> -C ₄ H ₉ (1a)	PhI (2a)	1.5/1	13	80 (3a)
2	CH ₃	H (1b)	PhI (2a)	1.5/1	23	82 (3b)
3	C ₂ H ₅	H (1c)	PhI (2a)	1.5/1	19	77 (3c)
4	(CH ₂) ₄	H (1d)	PhI (2a)	1.5/1	11	85 (3d)
5	(CH ₂) ₅	H (1e)	PhI (2a)	1.5/1	19	77 (3e)
6	H	Ph (1f)	PhI (2a)	1.5/1	18	44 (3f)
7	H	<i>n</i> -C ₄ H ₉ (1a)	<i>p</i> -MeOC ₆ H ₄ I (2b)	1.2/1	11	65 (3g)
8	H	<i>n</i> -C ₄ H ₉ (1a)	<i>p</i> -CH ₃ C ₆ H ₄ I (2c)	1.5/1	11	65 (3h)
9	H	<i>n</i> -C ₄ H ₉ (1a)	<i>p</i> -Me ₂ NCOC ₆ H ₄ I (2d)	1/1.5	11	62 (3i)
10	H	<i>n</i> -C ₄ H ₉ (1a)	<i>p</i> -CH ₃ COC ₆ H ₄ I (2e)	1/1.2	46	75 (3j)
11	H	<i>n</i> -C ₄ H ₉ (1a)	<i>p</i> -NCC ₆ H ₄ I (2f)	1/2	60	82 (3k)
12	H	<i>n</i> -C ₄ H ₉ (1a)	<i>p</i> -MeO ₂ CC ₆ H ₄ I (2g)	1/2	24	85 (3l)
13	H	<i>n</i> -C ₄ H ₉ (1a)	2-thienyl bromide (2h)	1.5/1	21	66 (3m)

^a Reaction was carried out using 0.3–0.5 mmol of 3,4-alkadienoates. ^b Isolated yield.

is <4%, if any. Even with ethyl 2-methyl-3,4-pentadienoate **1g**, the reaction afforded ethyl 2(*E*),4-pentadienoate **3n** highly stereoselectively, which was established by the NOE study of **3n** and **4n** (Scheme 2). The stereochemistry may be

In conclusion, we have established a highly regio- and stereoselective protocol for the synthesis of 2(*E*),4-alkadienoates. Due to the high stereoselectivity and easy availability of the starting allenes,¹⁰ this method will be useful in organic synthesis. Further studies in the area are being pursued in our laboratory.

Scheme 2

explained by the favorable *trans* orientation of the 3-vinyl group and the ester group referring to the α,β-C=C bond.

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Supporting Information Available: Analytical data for the compounds prepared, ¹H NMR and ¹³C NMR spectra of those compounds (PDF). These material is available free of charge via the Internet at <http://pubs.acs.org>.

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